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ABSTRACT

Leakage currents through insulators have received continuous attention for several decades, owing to their importance in a wide range of technologies and interest in their fundamental mechanisms. This work investigates leakage currents through atomic layer deposited Al₂O₃ grown on SrTiO₃. This combination is not only a key building block for oxide electronics but also a clean system for studying the leakage mechanisms without interfacial layers that form on most of the conventional bottom electrodes. We show how tiny differences in the deposition process can have a dramatic effect on the leakage behavior. A detailed analysis of the leakage behavior rules out Fowler-Nordheim tunneling (FNT) and thermionic emission. We conclude that the conduction mechanism is trap-related, and we ascribe it to trap-assisted tunneling or to Poole-Frenkel mechanisms. After annealing the sample in air, currents are reduced, which is ascribed to the transition from a trap-based mechanism to FNT, due to the elimination of the traps. The dramatic role of the assumptions regarding the flatband voltage used for analysis is critically discussed, and the sensitivity of the extracted parameters to this magnitude is quantitatively described. We show that future field-effect devices based on structures similar to those described here should be able to modulate $>10^{13}$ electrons/cm² in their channels. These results demonstrate ideas for reducing and analyzing leakage currents in insulators and highlight some of the possible approaches and pitfalls in their analysis, stressing the importance of the flatband voltage on the extracted parameters.

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I. INTRODUCTION

Alumina (Al₂O₃) grown by atomic layer deposition (ALD) is a widespread insulating oxide. The motivation for studying this material ranges from understanding various fundamental physical aspects of Al₂O₃^{1,2} to its considerable potential for applications in electronics, optics, and many other fields. Some two decades ago, ALD Al₂O₃ received considerable attention as a potential gate insulator for Si technology,³ owing to its large bandgap. Despite the preliminary interest, Al₂O₃ was eventually sidelined by Hf-based oxides as the high-k gate insulator for ultrascaled Si logic devices.⁴ Nonetheless, Al₂O₃ has found other uses in Si technology, such as in ultrathin layers for effective work function adjustment.^{5,6} Beyond Si technologies, ALD Al₂O₃ emerged as the best passivation layer for Ge,⁷ high-Ge content SiGe devices,⁸ and with III-V-based devices as well.⁹ Other back-end micro-electronic applications of ALD-Al₂O₃ include metal-insulator-metal (MIM) capacitors for resistive switching random-access memory (RRAM) devices,¹⁰ antifuse devices,¹¹ and others.¹²

More recently, the wide bandgap semiconductor β -Ga₂O₃ has emerged as a promising candidate for power devices.¹³ In such

roles, the ability to apply high fields is critical for the performance of power field-effect devices, and here, the high bandgap of Al₂O₃ is attractive in reducing the gate leakage currents.^{14,15}

Another potential application of ALD-Al₂O₃ is in oxide electronics, where oxide materials are harnessed for functionalities beyond their insulating properties. This field was greatly invigorated by the discovery of a 2D electron gas (2DEG) at the interface between some insulating oxides.^{16,17} One of the promising applications is an oxide field-effect transistor (FET), which utilizes the 2DEG as a confined electron channel.¹⁸ Such devices were demonstrated by epitaxially growing LaAlO₃ (LAO) on top of single crystal SrTiO₃ (STO) substrates.^{19–21} The subsequent discovery of oxide 2DEG based on amorphous oxides grown on STO²² has quickly paved the way to the application of ALD-Al₂O₃ for this purpose as well.²³ This concept has been extended by the replacement of STO crystals by thin TiO₂ layers, also grown by ALD.²⁴ This considerably increases the scalability of oxide electronics, by circumventing the use of single crystalline STO substrates,²⁵ available in limited sizes. Similar Al₂O₃/STO and Al₂O₃/TiO₂ structures

have also been suggested as selectors for memristor crossbar arrays,²⁶ for gas sensors,²⁷ spintronic devices,^{28,29} and field-effect transistors.²⁵

The wide bandgap of STO of 3.2 eV ²³ makes leakage reduction even more challenging, owing to the relatively low possible barriers with the insulator. We previously addressed this issue spectroscopically by investigating the band alignment at the $\text{Al}_2\text{O}_3/\text{STO}$ interfaces and reported barriers of $2.0 \pm 0.3\text{ eV}$ and $1.4 \pm 0.2\text{ eV}$ for electrons and holes, respectively.³⁰

Al_2O_3 growth is the easiest and most widespread ALD process, and it can be robustly performed over a wide range of temperatures and other process conditions³³ with excellent results. While many applications are relatively insensitive to the growth conditions, thin gate insulators can be extremely dependent on growth parameters. Otherwise-excellent Al_2O_3 films may exhibit high leakage currents, poor reproducibility, reliability problems, and other issues.

Considering their importance, leakage currents through ALD Al_2O_3 have been an integral part of its development since their early days on Si.³ Understanding the fundamental properties of the leakage currents can be done by studying leakage through ALD- Al_2O_3 grown on a semiconductor, on metal-coated substrates, or on conductive oxides. However, interfacial layers are typically formed at oxide/semiconductor interfaces, which complicate the analysis of leakage through the Al_2O_3 layer.³ Many metals form a native oxide surface layer, and when used as the back electrode, this layer adds an additional insulator in series; metals that do not have surface oxides, such as Pt, are typically problematic for nucleation of many ALD oxides due to their surface chemistry, which can result in lower quality films.

Conductive oxides are, therefore, more suitable substrates for studying leakage through ALD- Al_2O_3 . With $\beta\text{-Ga}_2\text{O}_3$ devices, the oxide substrate is already part of the device, and this issue has been addressed by several works, for example, Hung *et al.*¹⁴ and Bhuiyan *et al.*¹⁵ who both reported trap-assisted tunneling as the dominant Al_2O_3 leakage mechanism and extracted a trap energy of 1.1 eV below the conduction band. Another conductive oxide back electrode, indium tin oxide (ITO), has been employed by Spahr *et al.*³⁴ who reported a thorough investigation of the leakage currents through low-temperature ALD- Al_2O_3 grown at 80°C . While low temperature processes are crucially important for some applications, their resulting stoichiometry can be less ideal than films grown at $200\text{--}300^\circ\text{C}$, which is important for leakage reduction.

In this work, we address the leakage currents of ALD- Al_2O_3 grown on conductive STO substrates. We briefly demonstrate that a default ALD recipe is far from ideal for this task, and by comparison with a more optimized process, we obtain further insight into the conduction process. The motivation for this study is twofold: to evaluate ALD- Al_2O_3 and its limits for STO-based oxide electronics, and by employing STO as a conductive back electrode, we aim to understand the leakage mechanisms through Al_2O_3 , a question that is applicable to many technologies beyond oxide electronics.

II. EXPERIMENTAL

(001) 0.01 wt. % Nb-doped STO (Nb:STO) crystals (CrysTec GmbH) were TiO_2 terminated using the “extended Arkansas” method.³⁵ This process started with solvent sonication cleaning (acetone and methanol), followed by a 3:1 HCl- HNO_3 treatment. A two-step anneal was performed, starting with 1000°C for 1 h in air

and followed by 650°C for 30 min in flowing O_2 . A 10 nm thick amorphous Al_2O_3 layer was grown by ALD (Ultratech/Cambridge Nanotech Fiji G2) using trimethyl-aluminum (TMA) and water as the precursors. Two recipes were used and compared: recipe A is based on the manufacturer’s default recipe for Al_2O_3 , was performed at a substrate temperature of 300°C (0.06 s precursor pulses and 4 s purge pulses), and recipe B was optimized by extending the water pulse by a factor of 5, the TMA purging by $\times 2.5$ and the water purging by $\times 1.25$, at a substrate temperature of 280°C . A film thickness of $10 \pm 0.5\text{ nm}$ was confirmed by x-ray reflectivity (XRR, acquired using a Rigaku SmartLab and analyzed with GlobalFit 2.0). 50 nm Pt pads were deposited through a shadow mask using e-beam evaporation and a back contact was prepared by e-beam evaporation of a 300 nm blanket Al on the back of the wafer. Current density-voltage (I - V) and capacitance-voltage (C - V) characteristics were measured using a Keithley SourceMeter 2450 instrument and a Keysight E4980A precision LCR meter, respectively, in a shielded light-sealed box with a home-built heating stage. After measuring the sample, it was annealed in air in a tube furnace for 30 min at 500°C (measured on the outer tube surface). We note that under certain conditions, both amorphous²³ and crystalline^{36,37} Al_2O_3 grown on STO can cause the formation of conductive interfaces. However, with the process conditions employed here, the interface remains insulating. This was confirmed by the deposition of Al contacts on Al_2O_3 , grown on undoped STO and performing 4-point resistivity measurements; these showed that whatever conductivity present at the interface was below the measurement limit ($<0.2\ \mu\text{S}$). In addition, x-ray photoelectron spectroscopy (XPS) of this interface, carried out a through 4 nm layer of Al_2O_3 , did not show any Ti^{+3} , which is a fingerprint of interface conductivity and oxygen vacancies. These results are reported elsewhere.³⁰

III. RESULTS

The Pt/ Al_2O_3 /Nb:STO structures are treated as metal insulator semiconductor (MIS) capacitors, where, in the general case, the applied voltage on the gate (V_g) can be expressed by the following:³⁸

$$V_g = V_{FB} + V_{ox} + \psi_s, \quad (1)$$

where V_{FB} is the flatband voltage, V_{ox} is the voltage drop across the oxide (Al_2O_3), and ψ_s is the band bending, or surface potential of Nb:STO. Since Moon *et al.*³⁹ calculated significant ψ_s values in 0.7% doped samples, here C - V measurements were conducted at a frequency range of $5\text{--}800\text{ kHz}$ to assess the possibility of depletion in Nb:STO. As the doping here is $\times 70$ lower, one would expect higher ψ_s values; however, these were found to be negligible as detailed below. C - V measurements were conducted in the voltage ranges of -4 V to 4 V where leakage is undetectable [Fig. 2(a)], in order to ensure our accurate interpretation of the capacitance. Figure 1 shows little voltage or frequency dependence of the capacitance. No significant capacitance reduction is observed with voltage, indicating that ψ_s is independent of V_g at the measured voltage range. Fermi level pinning and other effects can result in some constant ψ_s , but this contribution to Eq. (1) can be included in V_{FB} instead. We therefore conclude that ψ_s is insignificant in Eq. (1), leaving V_{FB} as the major missing piece for relating between the applied gate voltage and V_{ox} . For the initial estimation,

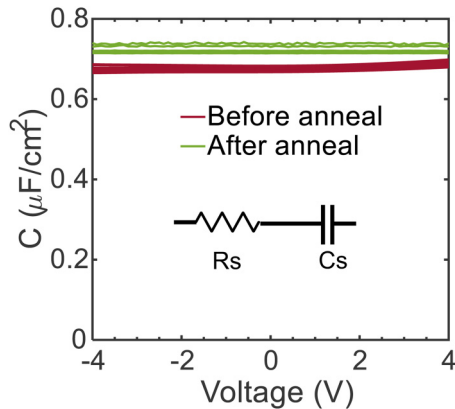


FIG. 1. CV measurements for 0.01 wt. % Nb:STO grown with recipe B using the C_s - R_s model (inset) in the frequency range of 5–800 kHz, before and after thermal anneal.

the flatband voltage is roughly approximated as $V_{FB} = \phi_{Pt} - \phi_{STO} = 1.45$ V, where ϕ_{Pt} is the Pt effective work function that was measured as 5.35 V on Al_2O_3 ⁴⁰ and ϕ_{STO} is approximated as the STO electron affinity of 3.9 V.⁴¹ Additional possible contributions to V_{FB} are neglected at this point, an assumption to be revisited later. Therefore, the electric field in the oxide, E , is estimated as

$$E = V_{ox}/d = (V_g - V_{FB})/d, \quad (2)$$

where d is the insulator thickness. In the absence of Nb:STO depletion, the Al_2O_3 permittivity can be extracted directly from the CV plot using the parallel plate capacitor expression $C_{ox}/A = \epsilon_0 \epsilon_r / t_{ox}$, where ϵ_0 is the vacuum permittivity, ϵ_r is the relative permittivity of Al_2O_3 , and A is the pad area.

Capacitance was extracted from the complex impedance using the C_s - R_s model⁴² (Fig. 1, inset), where C_s is the series capacitance and R_s is the series resistance, acquired at low and high frequencies, respectively. At accumulation (+4 V), the extracted C_s at 5 kHz and R_s at 800 kHz before thermal anneal are $0.68 \mu F/cm^2$ and 32Ω , respectively, while after anneal, the extracted C_s at 5 kHz and R_s at

800 kHz are $0.74 \mu F/cm^2$ and 94Ω , respectively (slight oxidation of the surface of the Al back contact may account for this small R_s increase). Relative permittivity values of 7.7 and 8.4 were extracted before and after thermal anneal, respectively, in agreement with previous reports.^{41,43} The small increase in the capacitance can be ascribed to the slight densification, which could increase the permittivity⁴⁴ and reduce the thickness.

We start the leakage analysis by comparing ALD recipe A (based on the instrument default) with recipe B (an optimized recipe). The JV behavior [Fig. 2(a)] shows that the default recipe A exhibits considerably higher leakage currents at lower voltages. In the most extreme cases, recipe A has over 2 orders of magnitude higher leakage currents at the same voltage compared to the optimized recipe B (e.g., around 4 V). Altogether, our optimized recipe yields undetectable leakage ($<10^{-8}$ A/cm²) at fields of under ± 4 MV/cm, and no signs of breakdown below 5 MV/cm in the devices tested here. In comparison, Moon *et al.* demonstrated 4 nm Al_2O_3 /Nb:STO heterojunction and observed leakage of above 10^{-3} A/cm² at an approximated electric field of 2.5 MV/cm⁻¹ (Ref. 39); however, at these ultrathin dimensions, additional mechanisms may come into play. These results illustrate how seemingly identical Al_2O_3 layers grown under similar conditions can vary wildly in their performance as gate insulators, as a result of minute process details.

From this point onward, all analysis is done on recipe B samples. Beyond the measurement voltage range, devices were found to be prone to irreversible damage. The relevant region for positive bias leakage was analyzed at varying temperatures. Figure 2(b) exhibits distinct temperature dependence, and a current increase of $\times 2$ was measured by increasing the temperature from 22 to 70 °C. The sample was measured once again after anneal (500 °C for 30 min in air) and showed two main noticeable differences: for any given voltage value, the leakage is noticeably smaller, and more importantly, the temperature dependence was diminished, to be addressed later.

Four possible leakage mechanisms were considered: Fowler-Nordheim tunneling (FNT), trap-assisted tunneling (TAT), Poole-Frenkel (PF) emission, and Schottky-Richardson thermionic emission (TE). The temperature dependence contradicts the FNT mechanism, where no temperature dependence is expected within its most simplified picture. On the other hand, PF, TE, and TAT models qualitatively agree with the measured temperature trend.

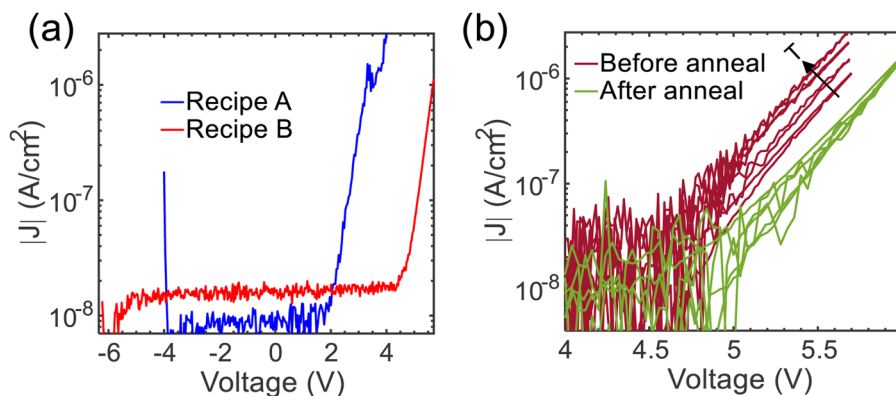


FIG. 2. (a) JV comparison between recipe A and recipe B. The measured noise floor of up to 4.5 pA at 22 °C corresponds to 2×10^{-8} A/cm². (b) Varying-temperature JV taken at the positive bias leakage region of recipe B, before and after annealing. Before annealing, the temperature ranges from 22 to 70 °C, and after annealing, the range is 22–50 °C.

The positive bias of the JV curve (5–5.7 V) was fitted by the above-mentioned mechanisms (Fig. 3). The relationship between the current density and the electric field in the PF emission model is given by⁴⁵

$$J \propto E \exp\left[-q\left(\phi_t - \sqrt{qE/(\pi\epsilon_0\epsilon_r)}\right)/kT\right], \quad (3)$$

where q is the electron charge, ϕ_t is the trap energy level below the Al_2O_3 conduction band, ϵ_r is the Al_2O_3 (high-frequency) relative permittivity, and k is the Boltzmann constant. The averaged relative permittivity is extracted from the slope of the linear fit of PF [Fig. 3(b)], determined to be 3.4 ± 0.5 , which is somewhat higher than the extracted value from optical measurements⁴⁶ of 2.2–2.6.^{47,48} Furthermore, other works suggested that PF has a negligible probability to be a dominant mechanism.⁴⁹

However, this analysis initially assumed a value of 1.45 V as the flatband voltage, an approximation of the effective work function difference, neglecting all other potential parasitics such as fixed oxide charges that are likely to affect the flatband voltage.⁴⁵ Without knowledge of V_{FB} , one cannot link V_g to the field [Eq. (2)], which can introduce considerable errors to the physical parameters extracted from the conduction mechanisms. The constant $C(V)$ behavior here makes V_{FB} extraction challenging and requires many assumptions. Therefore, a range of flatband voltages was considered in order to examine the effect of this elusive magnitude on the parameters extracted from each model. Analyzing the PF conduction mechanism with a broader range of flatband voltages, and considering the high-frequency permittivity values described above, yields reasonable flatband range voltages of -0.3 to -1.3 V, in which PF emission is possible [Fig. 3(c)] and thus cannot be eliminated.

The relationship between the current density and the electric field in the TE model is given by⁴⁵

$$J = A^* T^2 \exp\left[-q\left(\phi_B - \sqrt{qE/(4\pi\epsilon_0\epsilon_r)}\right)/kT\right], \quad (4)$$

where ϕ_B is the energy barrier between the conduction band edges of Nb:STO and Al_2O_3 , and A^* is the effective Richardson constant. An average energy barrier of 1.7 ± 0.1 eV was extracted from the intercept of the linear fit of TE [Fig. 3(d)], which is in reasonable agreement with the barrier of 2.3 ± 0.3 eV, obtained by spectroscopy.³⁰ However, the averaged relative permittivity of 0.8 ± 0.1 , extracted from the slope, does not agree with the reported value of ~ 2.4 .⁴⁷ While a good agreement can be obtained for the barrier ϕ_B , no flatband voltage in the (arbitrary) range examined could provide a physically relevant value for the permittivity [Fig. 3(f)], which eliminates the TE mechanism.

Next, we consider the validity of the TAT mechanism, where the relationship between the current density and the electric field, according to Fleischer *et al.*,⁵⁰ can be simplified by the following expression:

$$J = 2C_t N_t q \phi_t \times (3E)^{-1} \exp(-A\phi_t^{-3/2} E^{-1}), \quad (5)$$

where C_t is a slowly varying function of electron energy,⁵¹ N_t is the trap density, and $A = 4\sqrt{2}qm^* \times (3\hbar)^{-1}$, where m^* is the electron's

effective mass in Al_2O_3 and \hbar is the reduced Planck constant. The value used for m^* is $0.23m_0$ (Ref. 3). This specific TAT model was chosen for its simplicity and its ability to analyze the physical parameters, but we note that TAT modeling has many other variants.

By plotting $\ln(J/E)$ as a function of E^{-1} , we extract an average trap energy level of 1.6 eV below the conduction band edge of Al_2O_3 [Fig. 3(h)]. Theoretical analyses of oxygen vacancies in Al_2O_3 ^{31,32,52} predict typical energy levels of 2 eV below the conduction band, in some agreement with this experimental observation. After extracting these parameters, the accuracy of this procedure is examined by simulating the TAT integral $J = \int_0^{X_1} qC_t N_t [P_1 P_2 / (P_1 + P_2)] dx$ where $X_1 = (V_g - \phi_t)/E$ and P_1 and P_2 are the tunneling probabilities.⁵¹ This simulation is presented in Fig. 3(h) and implies that this simplified model is a good approximation for the TAT integral. TAT provides possible values for the trap levels that vary slowly with V_{FB} , and none can be ruled out. The trap density N_t is seen to vary wildly with the assumed V_{FB} [Fig. 3(i)]; changes of ~ 1 V in the former result in two orders of magnitude difference in the latter. We, therefore, conclude that extracting the trap density from this model is unreliable in the absence of precise knowledge of the flatband voltage and that TAT remains a likely candidate. While the basic TAT models are temperature-independent, some temperature dependence has been observed with TAT³⁹ and modeled by Yu *et al.*⁵³ who accounted for the Fermi-Dirac distribution at the injecting electrode; however, simulation using this model could not yield a satisfying fit for all data as well. We have, therefore, ruled out TE and cannot rule out TAT and PF at this point.

Last, we examine FNT, where the relationship between the current density and the electric field is given by the following expression:⁴⁵

$$J \propto E^2 \exp\left[-4\sqrt{2}qm^* \phi_B^{3/2} \times (3\hbar E)^{-1}\right]. \quad (6)$$

An FNT linear fit is shown in Fig. 3(k). An average barrier height of 1.4 V is extracted from the linear slopes of $\ln(J \times E^{-2})$ vs E^{-1} plots, which does not agree with the spectroscopically measured 2.3 ± 0.3 V.³⁰ The flatband analysis [Fig. 3(l)] yields a good agreement to the expected barrier value of 2.3 eV at low flatband voltages. While this may appear as a success, the increase of the currents with the temperature does not agree with the basic FNT model. Modifications to FNT can incorporate temperature dependence,⁵⁴ but FNT should produce the lowest currents from all the mechanisms discussed above; the observation of lower leakage currents after annealing, therefore, strongly indicates a trap-based conduction mechanism (either PF or TAT) as the leakage mechanism for the unannealed sample. As a result of the large energy barrier between STO and Al_2O_3 , it can be safely assumed that the first step of the dominant conduction mechanism, whatever it is, is tunneling electrons from STO to traps inside Al_2O_3 . Subsequently, the electrons either gain enough thermal energy to escape out to the conduction band (PF) as modeled by Jeong *et al.*⁵⁵ in their so-called “tunnel assisted PF” (TAPF) or tunnel to the other electrode, as was indicated by Yu *et al.*⁵³

Since the slopes of the JV curves before and after annealing are similar, repeating the analysis of Fig. 3 on the annealed samples produces nearly identical parameters. However, the disappearance

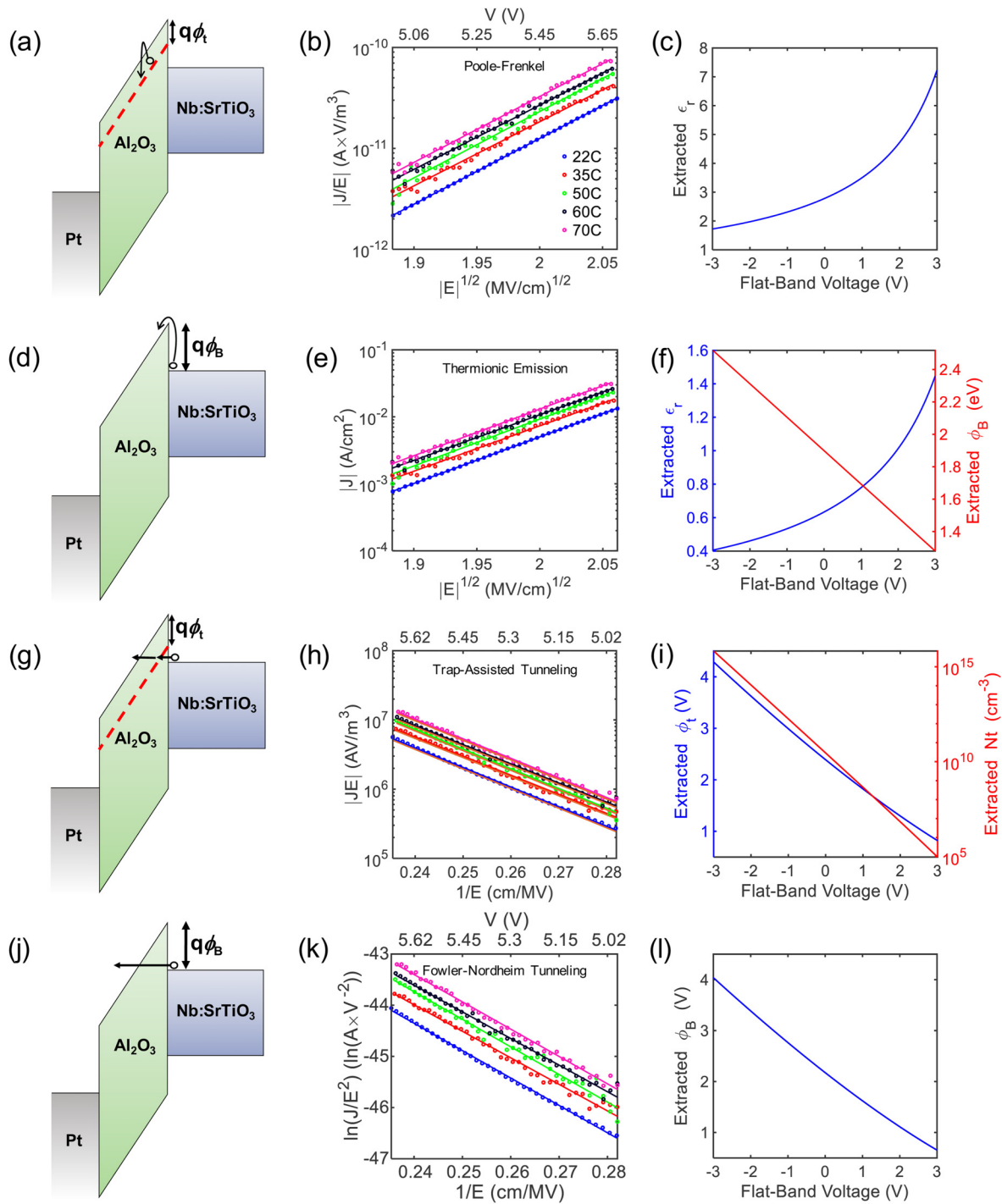


FIG. 3. Conduction mechanisms analysis at positive bias. (a)–(c) Poole-Frenkel (PF), (d)–(f) thermionic emission (TE), (g)–(i) trap-assisted tunneling (TAT), and (j)–(l) Fowler-Nordheim tunneling (FNT). The first panel of each row shows a schematic of the conduction mechanism. The second panel presents the temperature-dependent fits to the model represented in that row, assuming $V_{FB} = 1.45$ V. Symbols represent the measured data points and lines represent their linear fits. An additional orange line in panel (h) represents the simulated TAT curves (which coincide with the fits, see text). The third panel of each row illustrates the influence of the flatband voltage assumption on the parameters extracted from that model.

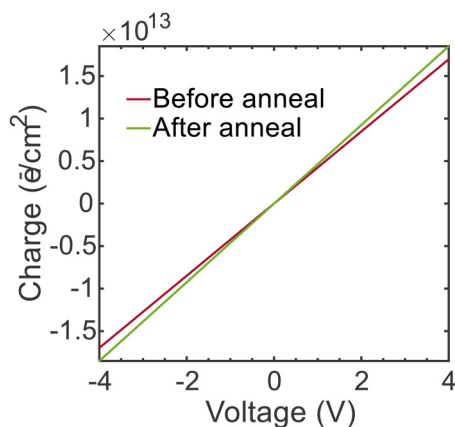


FIG. 4. $Q = CV$ analysis for annealed and unannealed structures, showing the possible degree of charge modulation using the studied MIS structures.

of the temperature dependence following annealing and the overall leakage reduction favor FNT. Since our interpretation of TAT or PF in the unannealed samples ascribed the enhanced electron conductivity to the presence of oxygen vacancies,^{39,52} it is quite likely that annealing in air oxidized the sample enough to decrease the trap density and hence minimize the contribution of the traps, leaving FNT as the dominant current mechanism.

For oxide electronics, for example, field-effect transistors (FETs) with 2DEG channels, depleting the 2DEG is necessary for closing this “normally on” device.⁵⁶ To maximize the on/off channel resistivity ratio in such devices, it is desirable to have many electrons in the “on” state, but not too many so that it could be effectively switched off completely in the “off” state. This makes the charge modulation an important parameter of the gate stack for optimizing the on/off ratio, a key performance metric for transistors (we note that in nanoscale devices, other issues can strongly affect this magnitude³⁸). We therefore estimate the potential for charge modulation $\Delta Q = C_{\text{Al}_2\text{O}_3}/\Delta V$. For the unannealed structure, we obtain a modulation of $\sim 4 \times 10^{12}$ electrons/cm² per 1 V on the gate (0.45 MV/cm using $V_{\text{FB}} = 1.45$ V) or a modulation of $\sim 1.7 \times 10^{13}$ electrons/cm² at gate voltages of up to 4 V (2.55 MV/cm), before the onset of detectable leakage currents (Fig. 4). An additional higher- k layer may be considered for creating a bilayer structure, a known strategy for reducing the field on the Al_2O_3 without significant reduction of the capacitance.⁵⁷ Altogether, our data provide design guidelines for an oxide FET, and particularly for engineering its 2DEG properties, toward achieving low off-state currents and maximizing their $I_{\text{on}}/I_{\text{off}}$ ratios.

IV. CONCLUSIONS

We show that while ALD Al_2O_3 is a mature, well-established process for gate insulator applications, optimization of small process details can lead to huge benefits in mitigating leakage. The optimized structures can be useful as gate stacks for oxide electronics, owing to their low leakage that enables significant charge modulation of an underlying channel.

Analysis of the conduction mechanisms revealed that while all those considered provide a good mathematical agreement to the data, careful consideration of the extracted parameters rules out TE and FNT in the unannealed samples. Moreover, we show that the effect of the assumed flatband voltage on the extracted parameters can be huge, necessitating careful handling of this parameter. Altogether, we identify trap-based conduction (TAT or PF) as the leakage mechanism for unannealed samples. These traps are ascribed to oxygen vacancies in Al_2O_3 . A moderate anneal reduces the leakage currents and alleviates their temperature dependence. This observation points to Fowler-Nordheim tunneling (FNT) as the leakage mechanism, indicating the possible mitigation of the oxygen vacancies during annealing.

SUPPLEMENTARY MATERIAL

The data used to generate all the figures in this work are available in the online [supplemental material](#).

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